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Sensitive determination of a novel bisphosphonate, YM529, in plasma, urine and bone by high-performance liquid chromatography with fluorescence detection

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Abstract

A high-performance liquid chromatographic method for the sensitive determination of 1-hydroxy-2-(imidazo[1,2-a]pyridin-3-yl)ethane-l,l-bisphosphonic acid monohydrate (YM529) in plasma, urine and bone is described. Plasma obtained in high-dose animal studies is treated by method A, a simple method using 1 ml of plasma, which is based on deproteinization of plasma followed by coprecipitation of the drug with calcium phosphate and dissolution of the precipitate in EDTA. Plasma obtained in low-dose clinical studies is treated by method B, a more sensitive method using 4 ml of plasma, which is based on direct precipitation of the drug prior to the deproteinization in method A. Urine and bone samples are prepared by solid-phase extraction using a Sep-Pak C_{18} cartridge coupled with method A. The drug is separated with a reversed-phase column using a mobile phase at pH 7, and detected with a fluorescence detector following postcolumn alkalization of the mobile phase to enhance fluorescence intensity. The limit of determination is 0.2 ng/ml for method A and 0.05 ng/ml for method B in plasma, 0.05 ng/ml in urine, and 5 ng/g in bone.

1. Introduction

1-Hydroxy-2-(imidazo[1,2-a]pyridin.3_yl)eth_ ane-1,1-bisphosphonic acid monohydrate (YM529, I) (Fig. 1) is a newly synthesized bisphosphonate which is expected to prove clinically useful in the treatment of osteoporosis and hypercalcemia. In animal models, this agent shows a 100-fold greater potency against bone resorption than pamidronate, a second genera-

Fig. 1. Chemical structures of I (YM529) (a) and the internal standard (b).

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tion bisphosphonate [1], and is also 10 times more effective than disodium dihydrogen(cycloheptylamino)methylenebisphosphonate monohydrate (YM175, II), our previously developed bisphosphonate compound [1,2].

We previously reported an assay method for II by HPLC with electrochemical detection [3]. The limit of determination of II was 5 and 0.5 ng/ml in plasma, using samples of 1 and 10 ml, respectively, and 1 ng/ml in urine and 25 ng/g in bone. The determination of I in plasma and urine, however, was anticipated to require a 10-times higher sensitivity than that for II, because effective doses of I are as low as 1/10 of those of II. Although II lacks suitable functional groups for UV detection and derivatization, compound I possesses an imidazopyridin ring which emits intense fluorescence, especially at alkaline pH. We developed a sensitive assay method for I in plasma, urine and bone using reversed-phase HPLC with fluorescence detection.

2. Experimental

2.1. Chemicals and reagents

YM529 (I) and the internal standard (I.S.), 1-hydroxy- 2 -(imidazo[1,2-a]pyridin- 3 -yl)propane-l,l-bisphosphonic acid (Fig. 1), were synthesized at our Central Research Laboratories (Tokyo, Japan). Calcium phosphate (tribasic) was purchased from Nacalai Tesque (Kyoto, Japan) and used as a solution of 5 mg/ml in 0.1 M HCI. Distilled water was used for all procedures.

2.2. Standard solution

Compound I and the I.S. were each dissolved in a NaOH solution at two equivalents of these compounds, then diluted with water to a concentration of 1 mg/ml. These solutions were stored at 4°C for use as stock solutions. Standard solutions of I and the I.S. were prepared by diluting the stock solutions with water to suitable concentrations. Polypropylene tubes were used for the dilution because the studied compounds are readily adsorbed onto a glass surface in water at low concentrations.

2.3. Chromatographic conditions

The HPLC system consisted of a 655A-11 pump coupled with an L-5000 LC controller (Hitachi, Tokyo, Japan), a WISP712 autosampier (Waters Assoc., Milford, MA, USA), an RF-535 fluorescence detector (Shimadzu, Kyoto, Japan), a Develosil ODS-5 column (15×0.46) I.D. cm, Nomura Chemical, Tokyo, Japan), and a C-R4AX integrator (Shimadzu). In routine analysis, the column could used for 200-300 injections.

The mobile phase consisted of methanol -10 mM sodium pyrophosphate containing 1 mM tetrabutylammonium phosphate, adjusted to pH 7.0 by orthophosphoric acid $(5:95, v/v)$, and was delivered at a flow-rate of 1.0 ml/min. Column temperature was maintained at 40°C. The column eluate was mixed with 0.2 M NaOH using an LC-6A pump (Shimadzu) at a flow-rate of 0.1 ml/min, and then monitored fluorometrically at excitation and emission wavelengths of 281 and 391 nm, respectively.

2.4. Sample preparations

Plasma: method A

To 1 ml of plasma in a 10-ml polystyrene tube, 50 μ l of 0.1 μ g/ml I.S. solution was added. The sample was deproteinized by adding 4 ml of 6% (v/v) perchloric acid, and centrifuged at 2000 g for 10 min. The supematant was transferred to a 15-ml polypropylene tube, then spiked with 0.25 ml of calcium phosphate and 0.4 ml of 10 M NaOH. The mixture ($pH > 12$) was vortex-mixed for 30 s and centrifuged at 2000 g for 10 min to precipitate the analytes with calcium salts. The supematant was discarded. The precipitate was suspended in 4 ml of water. After centrifugation at 2000 g for 10 min, the precipitate was dissolved in 0.2 ml of 0.1 M ethylenediaminetetraacetic acid disodium salt (EDTA-2Na) in 0.2 M NaOH (pH $ca.$ 11). A $40-\mu l$ aliquot of the sample was injected onto the HPLC column. Compound I and the I.S. in the final solution were stable for at least 72 h at room temperature.

Plasma: method B

To 4 ml of plasma in a 15-ml polypropylene tube, 50 μ l of 0.2 μ g/ml I.S. solution, 4 ml of water, 0.1 ml of calcium phosphate and 0.1 ml of 10 M NaOH were added. The mixture ($pH > 12$) was vortex-mixed for 30 s and centrifuged at 2000 g for 10 min. The supernatant was discarded. The precipitate was suspended in 1 ml of 0.1 M HCl, then deproteinized by adding 2 ml of 6% perchloric acid. After centrifugation at 2000 g for 10 min, the supernatant was treated in the same way as that obtained after deproteinization in method A, except that no calcium phosphate was added.

Urine

To 4 ml of urine in a 10-ml polystylene tube, 50 μ 1 of 0.2 μ g/ml I.S. solution, 0.1 ml of 2.5 M calcium chloride and 0.05 ml of 10 M NaOH were added. The mixture $(pH > 12)$ was vortexmixed for 30 s and centrifuged at 2000 g for 10 min. The supernatant was discarded, and the precipitate was dissolved in 3 ml of 0.1 M HCI and then applied to a Sep-Pak C_{18} cartridge, which was activated with 5 ml of acetonitrile and water. The cartridge was washed with 10 ml of water, 5 ml of 50% (v/v) acetonitrile, and 5 ml of water. The analytes were eluted with 6 ml of 0.2 M NaOH. The eluate was treated in the same way as the supernatant after deproteinization in method A, except that no $10 M$ NaOH was added.

Bone

Bone (250-500 mg) was digested in 4.7 volumes of conc. HCI for *ca.* 2 h at 50°C in a 10-ml capped glass tube. To the digest, 4.3 volumes of 10 M NaOH were added to form a transparent liquid phase. To a 15-ml polypropylene tube, 0.5 ml of the liquid phase (pH *ca.* 1), corresponding to 50 mg of bone, was transferred and spiked with 50 μ l of 0.4 μ g/ml I.S. solution and 3 ml of water, and then applied to a Sep-Pak C_{18} cartridge. The cartridge was washed with 10 ml of water and the analytes were eluted with 6 ml of 0.2 M NaOH. The eluate was spiked with 0.25 ml of calcium phosphate, vortex-mixed for 30 s, and centrifuged at 2000 g for 10 min. The precipitate obtained was dissolved in 4 ml of 0.05 M HCl and then spiked with 0.05 ml of 10 M NaOH. After mixing for 30 s and centrifugation at 2000 g for 10 min, the precipitate was dissolved in 0.1 M EDTA-2Na in 0.2 M NaOH, and 20 μ l of the sample was injected onto the HPLC column.

2.5. Preparation of calibration curves

A 50- μ l portion of each standard solution of I was added to 1 ml of plasma, 4 ml of plasma or urine, respectively. These spiked samples were processed as described above. Bone was spiked with 0.2 volume of a standard solution of I and processed as described above except that 4.6 and 4.2 volumes of conc. HCl and $10 M$ NaOH were added, respectively. Calibration curves were constructed by plotting the peak-height ratios of I to the I.S. *versus* the concentration of I.

3. Results

3.1. HPLC profiles

Typical chromatograms of plasma, urine and bone samples are shown in Figs. 2 and 3. Compound I and the I.S. were eluted at 6.2 and 9.2 min, respectively. The chromatograms obtained from control samples showed no interfering peaks at the retention times of I and the I.S. (data not shown).

3.2. Calibration curves and limit of determination

Calibration curves were linear $(r > 0.999)$ over the ranges 0.2-100 ng/ml for method A and $0.05-10$ ng/ml for method B in plasma, $0.05-10$ ng/ml in urine, and 5-2000 ng/g in bone. The limits of determination were 0.2 ng/ml for method A and 0.05 ng/ml for method B in plasma, 0.05 ng/ml in urine, and 5 ng/g in bone.

Fig. 2. Typical chromatograms of plasma samples containing I.S., treated by method A $[(a)$ and $(b)]$ and method B $[(c)$ and (d)]: (a) control dog plasma; (b) dog plasma obtained 1 h after oral administration of 1 mg/kg of I (4.01 ng/ml); (c) control human plasma; (d) control human plasma spiked with 0.5 ng/ml of I.

Fig. 3. Typical chromatograms of urine and bone samples containing I.S.: (a) control human urine; (b) control human urine spiked with 0.5 ng/ml of I; (c) control rat bone; (d) rat bone obtained 24 h after oral administration of 3 mg/kg of I (144 ng/g).

3.3. Recovery after sample preparation

Total recovery (mean \pm S.D., $n = 3$) of YM529 following sample preparation was $90.1 \pm 1.6\%$ for method A and $45.6 \pm 2.4\%$ for method B in plasma at 5 ng/ml, $62.3 \pm 3.5\%$ in urine at 5 ng/ml, and $59.4 \pm 0.8\%$ in bone at 100 ng/g.

3.4. Intra-assay accuracy and precision

As shown in Table 1, overall mean precision in method A, as defined by the coefficient of variation (C.V.), ranged from 1.5% to 13.6% for human plasma. Analytical accuracy, expressed as a percent difference of the mean observed value from the prepared concentration (Δ) , varied from 0% to +10.0%. In method B, C.V. and Δ values ranged from 3.9% to 10%, and from -2.0% to 2.0%, respectively.

C.V. and Δ values in assay of human urine ranged from 2.0% to 8.3%, and from -2.0% to 20%, and those for rat bone from 1.2% to 20%, and from -6.2% to 0%, respectively.

3.5. Application to the animal study

Fig. 4 shows plasma concentration of I determined by method A after oral administration at 0.3 mg/kg to dogs. Plasma concentration reached maximum levels at 2 h and could be detected up to 6 h after administration.

4. Discussion

Fluorescence of I in the alkaline mobile phase (pH *ca.* 11) was 4.5 times more intense than that in the mobile phase at pH 7. Although we tried reversed-phase HPLC using an alkaline mobile phase and several alkaline stable columns packed with polymer-based materials, sensitive analysis was not possible due to broadening of the peak. Therefore, compound I was separated using an octadecylsilica-based column with a neutral mobile phase, followed by mixing of the eluate with NaOH solution to enhance fluorescence intensity.

Sample preparation was based on coprecipita-

Sample	Prepared concentration $(ng/ml \text{ or } ng/g)$	Concentration found $(\text{mean} \pm S.D., n = 5)$ (ng/ml or ng/g)	Coefficient of variation (%)	Percentage difference"	
Plasma	0.2	0.22 ± 0.03	13.6	$+10$	
(method A)	2	2.02 ± 0.03	1.5	$+1.0$	
	20	± 0.40 20.0	2.0	0	
Plasma	0.05	0.050 ± 0.004	8.0	$\bf{0}$	
(method B)	0.5	0.51 ± 0.02	3.9	$+2.0$	
	5	4.9 ± 0.5	10	-2.0	
Urine	0.05	0.060 ± 0.005	8.3	$+20$	
	0.5	0.49 ± 0.02	4.1	-2.0	
	5	5.0 ± 0.1	2.0	Ω	
Bone	5	5.0 ±1.0	20.0	$\bf{0}$	
	50	46.9 ± 2.6	5.5	-6.2	
	500	495 ± 6	1.2	-1.0	

Table 1 Accuracy and precision for the determination of I in human plasma, human urine and rat bone

^a Percentage difference = $[(mean of concentration found - prepared concentration)/prepared concentration] \times 100$.

tion of the drug with calcium phosphate, in a way similar to the assay methods for other bisphosphonates [3-8]. This procedure involves a modification of the YM175 (II) assay [3], namely deproteinization of plasma, digestion of bone by HCI, and solid-phase extraction of urine and bone samples using a Sep-Pak C_{18} cartridge. Overall, the sample preparation in the assay for compound I is simpler than that for compound II, especially in the dissolution step of the final calcium precipitate. The precipitate obtained in

Fig. 4. Plasma concentration of unchanged drug after oral administration of I at 0.3 mg/kg to dogs, determined by method A. Each point represents the mean \pm S.E. of four dogs.

the assay for I is merely dissolved in EDTA solution, while that for II is treated in 4 steps, *e.g.* dissolution of the precipitate in phosphoric acid, removal of excess calcium ions by a cationexchange resin, filtration, and alkalization by NaOH.

We established two methods for the treatment of plasma samples, in the same manner as in the assay for II. Method A requires 1 ml of plasma, and is simpler and therefore useful for high-dose animal studies. Method B requires 4 ml of plasma, and is 4 times more sensitive than method A and therefore suitable for clinical studies at lower doses. Finally, the sensitivities in the assay of I are sufficiently higher than those for II, *i.e.* 25 times higher for method A and 10 times for method B in plasma, 20 times in urine, and 5 times in bone.

5. References

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